BANKURA ZILLA SARADAMANI MAHILA MAHAVIDYAPITH DEPARTMENT OF CHEMISTRY

<u>Programme Outcome, Programme Specific Outcome and Course Outcome</u> <u>For B.Sc. Honours (CBCS Pattern) in Chemistry</u> <u>2021-2022</u>

Department of Chemistry		After successful completion of three year degree program in
		Chemistry a student should be able to;
РО	Programme Outcomes	Description
PO 1	Sound domain	Acquiring sound knowledge of chemical concepts and
	knowledge	emerging issues in chemical science.
PO2	Academic and	To help the students in developing academic and scientific
	Scientific Endeavour	endeavour by fostering and nurturing the young talent for
		proper scientific pursuit.
PO3	Creative and Practical	Analysis of experimental data and their representation in the
100	Ability to analyse and	form of graphs and plots. Use of statistics as a means to
	deal with data	express complicated chemical data.
PO4	Familiarity with	Should be able to apply modern theories and approaches to
	Recent Developments	explain all spatial phenomena and relate nature with human
	in a Particular Field	inter relations
PO5	Environmental	Impact of environmental changes on human and how it can be
	Awareness	explained at a global and regional perspective.
PO 6	Laboratory Skill	The students are exposed to modern equipments in the
		Laboratory where they get hands-on training which help them
		to succeed at any entry-level position in chemical industry.
PSO	Programme Specific	Description
	Outcomes	
PSO 1	Critical appreciation of	Acquiring sound knowledge on the fundamentals of Physico-
	the Subject.	chemical concepts and applying them in practical and
		professional situations.
PSO 2	Academic and	To help the students in developing, Cultivating and
	Scientific Endeavour.	demonstrating the art of science learning and teaching by
		fostering and nurturing the young talent for proper scientific
		pursuit.
PSO 3	Scientific Attitude	Developing the right scientific temper compatible with
		creative impulse.
PSO 4	Technical Skill	Creating updated knowledge on research methodology and
	Development	developing skills in the application oriented Chemistry.

PSO 5	Environmental	Impact of environmental changes on human and its reflection	
	Consciousness	on society.	
PSO 6	Communication Skill	Classroom discussions, student seminar ,written assignments,	
		debates etc. help students to develop effective communication	
		skill which will aid them to enhance employability.	
PSO 7	Personality	Personality development skills are likely to help students in	
	Development	their professional and personal lives thus making them	
		responsible and sincere citizens of the socie \sqrt{ty} .	
PSO 8	Spirit of Team Work	Encouraging students to co-ordinate with one another in a	
		team environment rather than trying to excel individually.	
PSO 9	Basic Human Values	Study of various texts and mutual interaction among the	
		students inside and outside the class room help the learners to	
		understand human behavioural science.	
	<u>Course Outcomes</u>	B. Sc Chemistry (Honours Semester-1)	
Course		Outcomes	
		After completion of these courses students should be able	
		The completion of these courses students should be use	
CC-1 Organi	c Chemistry I	CO-1. To learn about bonding and physical properties of	
		organic molecules	
		CO-2. To learn general treatment of reaction mechanism	
		CO-3. To learn stereochemistry of organic molecules	
		CO-4 To learn separation techniques determination of	
		hoiling point and identification of organic	
		some syn de	
CC-2 Physic	al Chemistry I	CO-5. To learn properties and behaviors of gaseous state	
		CO-6. To learn chemical thermodynamics and its	
		application	
		CO-7. To learn kinetics of chemical reactions	
		CO-8. To study kinetics of chemical reactions	
		experimentally and determination of pH and	
		solubility product	
	Course Outcomes	B. Sc. Chemistry (Honours Semester-II)	
	Course Outcomes D. Se Chemistry (nonours Semester-11)		
Course		Outcomes	
		After completion of these courses students should be able	
CC-3 Inorganic Chemistry-I		CO-9. To learn about extranuclear structures of atoms	
		CO-10. To learn chemical periodicity	
		CO-11. To learn about acid base reactions. redox	
		reactions and precipitation reactions	
		CO-12. To learn redox titrations	

	(experimentally)
CC-4 Organic Chemistry-II	CO-13. To learn stereochemistry of organic molecules CO-14. To learn general treatment of reaction mechanism
	CO-15. To learn substitution and elimination reactions in organic chemistry
	CO-16. To prepare organic compounds, purify them and to determine melting point
<u>Course Outcomes B.</u>	<u>Sc_Chemistry (Honours Semester-III)</u>
Comme	Ontermore
Course	After completion of these courses students should be able
CC-5 Physical Chemistry-II	CO-17. To learn about transport process CO-18. To learn application of thermodynamics in chemistry
	CO-19. To learn about basic quantum mechanics CO-20. To determine viscosity, partition coefficient, equilibrium constant and to perform conductometric experiments
CC-6 Inorganic Chemistry-II	CO-21. To learn chemical bonding CO-22. To learn radioactivity and nuclear structure CO-23. To learn about iodometric and iodimetric titrations experimentally
CC-7 Organic Chemistry-III	 CO-24. To learn about chemistry of alkenes and alkynes CO-25. To learn aromatic substitutions CO-26. To learn about carbonyl compounds and organometallic reagents CO-27. To detect special elements and functional groups in organic compounds and to prepare suitable derivatives
SEC-1 Basic Analytical Chemistry	 CO-28. To learn about fundamental mathematical procedure and their applications in chemistry CO-29. To learn computer programming for statistical analysis CO-30. To handle numeric data Co-31. To learn about application of basic analytical procedures in chemistry
Course Outcomes	B. Sc Chemistry (Honours Semester-IV)
Course	Outcomes
	After completion of these courses students should be able

CC-8 Physical Chemistry-III	CO-32. To learn about transport process
	CO-33. To learn application of thermodynamics in
	chemistry
	CO-34. To learn about basic quantum mechanics
	CO-35. To determine viscosity, partition
	coefficient, equilibrium constant
	and to perform conductometric experiments
CC-9 Inorganic Chemistry	CO-36. To learn about general principle of metallurgy
	CO-37. To learn about chemistry of s and p block elements
	and also on inorganic polymers CO-3. To learn
	about basic coordination chemistry
	CO-38 To studies complexometric titration and inorganic
	preparation of complex salts
CC-10 Organic Chemistry	CO-39. To studies about nitrogen compounds and various
CC-10 Organic Chemisury	types of name reactions
	CO 40 To know about the rearrangement reaction and
	co-40 10 know about the realizingement reaction and
	stereochemical leachers of aliphatic and aromatic
	CO 41 To staling with a single staling single senthaling
	CO-41. To studies retrosynthesis analysis, ring synthesis,
	asymmetric synthessis
	CO-42. To learn about UV, IR and NMR spectroscopy
	CO-43 To learn about the estimation of organic compounds
	experimentally
SEC-2 Pharmaceutical Chemistry	CO-44. To learn about pharmaceuticals chemistry
	including drugs
	CO-45. To study about fermentation
Course Outcomes	B. Sc Chemistry (Honours Semester-V)
Course	Outcomes
	After completion of these courses students should be able
CC-11Inorganic Chemistry IV	CO-46 To learn about VBT and CFT magnetic colour
	noperties of coordination compounds
	CO_{17} To study coordination compounds
	CO_{48} To have idea about 2d 4d and 5d elements in term
	of algorranic configuration avidation states reday
	proportion apprending tion abamistry
	CO 40. To loom about the about internetions wetel
	CO-49. To learn about the chemistry of transitions metal
	and lanthanoids and actinoids

CC-12 Organic Chemistry V	CO-50. To learn about heterocyclic compounds and poly	
	nuclear aromatic compounds	
	CO-51. To study about alicyclic ompounds	
	CO-52 To understand about pericyclic reactions	
	CO-53. To know about the amino acids and	
	proteins	
	CO-54 To learn about nuclic acids	
DSE-1 Advanced Physical Chemistry	CO-55. To learn about Crystal Structure	
	CO-56. To learn about statistical thermodynamics	
	CO-57. To study about Specific heat of solid, 3rd law and	
	Adiabatic demagnetization	
	CO-58. To know about Computer Programming based on	
	numerical methods	
DSE-2 Green Chemistry	CO-59. To know about the principles of Green Chemistry	
	and Designing a chemical synthesis	
	CO-60. To study some examples of Green Synthesis/	
	Reactions	
	CO-61. To learn about Future Trends in Green Chemistry	
Course Outcomes B. Sc Chemistry (Honours Semester-VI)		
DSE -3 Analytical Methods in	CO-62: Understand the fundamental principles of	
Chemistry	qualitative and quantitative analysis.	
	CO-63: Apply classical methods like volumetric and	
	gravimetric analysis.	
	CO-64: Utilize instrumental techniques like spectroscopy	
	(UV-Vis, IR, AAS, etc.), chromatography (HPLC, GC),	
	and electrochemical methods.	
	CO-65: Interpret experimental data and draw meaningful	
	conclusions.	
	CO-66: Develop practical skills in using analytical	
	instruments and software.	
	CO-67: Understand the importance of quality control and	
	quality assurance in analytical chemistry.	
	CO-68: Apply analytical techniques to solve real-world	
	problems in various fields.	
DSE -4 Polymer Chemistry	CO-69. State the basic concept of polymer.	
	CO-70. Relate Tm, Tg and its significance.	
	CO-71. Apply the Polymerization techniques and Polymer	
	CO-72. Differentiate Natural and synthetic rubbers.	
	CO-/3. Distinguish Thermoplastic and thermosetting	
CC 12 In annual Classification V	resins.	
UC-13 Inorganic Chemistry V	CU-/4. Study different inorganic chemistry of different	
	biological process such as role of different elements	
	protoing nitrogen fixetion Distance theory	
	proteins, nitrogen iixation, Photosynthesis etc.	
	UU-15. Gain knowledge of organometallic compounds,	

	their use in catalysis.
	CO-76. Reaction kinetics and mechanism of reactions of
	coordination compounds.
	CO-77. Learn qualitative analysis mixture of inorganic salt
	mixture and determine their composition.
CC-14 Physical Chemistry IV	CO-78. Study different spectroscopic properties (UV,
	rotational, vibrational) of molecule to explain different
	molecular properties.
	CO-79
	To analyze different physicochemical behaviour of
	chemical compounds in respect of their interaction with
	light.
	CO-80.Learn to measure physicochemical data
	(absorbance, molar extinction coefficient, pH of buffer,
	CMC etc.) of some compounds and also their interaction
	with biomolecules using UV, IR spectrophotometer.

<u>Programme Outcome, Programme Specific Outcome and Course Outcome</u> <u>For B.Sc. Generic and Programme Cources (CBCS Pattern) in Chemistry</u> <u>2021-2022</u>

Department of Chemistry		After successful completion of three year degree program in
		Chemistry a student should be able to;
DO	D	Description
PO	Programme	Description
	Outcomes	
PO 1	Sound domain	Acquiring sound knowledge of chemical concepts and emerging
	knowledge	issues in chemical science.
DO1		
PO2	Academic and	To help the students in developing academic and scientific
	Scientific	endeavour by fostering and nurturing the young talent for proper
DO3	Endeavour Creative and	scientific pursuit.
103	Dreatical Ability to	of graphs and plots. Use of statistics as a means to express
	analyse and deal	complicated chemical data
	with data	complicated chemical data.
PO4	Familiarity with	Should be able to apply modern theories and approaches to
101	Recent	explain all spatial phenomena and relate nature with human inter
	Developments in a	relations
	Particular Field	
PO5	Environmental	Impact of environmental changes on human and how it can be
	Awareness	explained at a global and regional perspective.
PO 6	Laboratory Skill	The students are exposed to modern equipments in the Laboratory
		where they get hands-on training which help them to succeed at
DCO	D	any entry-level position in chemical industry.
PSO	Programme	Description
	Specific Outcomes	
PSO 1	Critical	Acquiring sound knowledge on the fundamentals of Physico-
	appreciation of the	chemical concepts and applying them in practical and professional
	Subject.	situations.
PSO 2	Academic and	To help the students in developing, Cultivating and
	Scientific	demonstrating the art of science learning and teaching by
	Endeavour.	fostering and nurturing the young talent for proper scientific
		pursuit.
PSO 3	Scientific Attitude	Developing the right scientific temper compatible with creative
		impuise.

PSO 4	Technical Skill Development	Creating updated knowledge on research methodology and developing skills in the application oriented Chemistry.
PSO 5	Environmental Consciousness	Impact of environmental changes on human and its reflection on society.
PSO 6	Communication	Classroom discussions, student seminar ,written assignments,
	Skill	debates etc. help students to develop effective communication skill which will aid them to enhance employability.
PSO 7	Personality Development	Personality development skills are likely to help students in their professional and personal lives thus making them responsible and sincere citizens of the socie \sqrt{ty} .
PSO 8	Spirit of Team Work	Encouraging students to co-ordinate with one another in a team environment rather than trying to excel individually.
PSO 9	Basic Human Values	Study of various texts and mutual interaction among the students inside and outside the class room help the learners to understand human behavioural science
Course	e Outcomes B. Sc C	hemistry Generic Elective and Programme (Semester-I)
Course		Outcomes After completion of these courses students should be able
GE-1/C-1A	Outcomes B. Sc Cl	CO-1. To learn about extranuclear structures of atoms CO-2. To learn chemical periodicity CO-3. To learn about acid base reactions, redox reactions and precipitation reactions CO-4. To learn redox reactions CO-5. To learn about fundamental features of organic chemistry CO-6. To learn about stereochemistry CO-7. To study about nucleophilic substitution and elimination reactions CO-8. To learn fundamentals features of alkanes, alkenes and alkynes
Course		Outcomes
		After completion of these courses students should be able

GE-2 /C-1B	CO-9. To learn about features of gaseous states
	CO-10 To learn properties of liquids
	CO-11. To learn about properties of solids
	CO-12 To learn kinetics of chemical reactions
	CO-13. To learn about chemical bonding and molecular
	structures
	CO-14. To learn about comparative study of n- block elements
	CO-15. To study about determination of physical and chemical
	narameters experimentally
	CO-16. To learn qualitative semimicro inorganic analysis
Course Outcomes B. Sc Ch	emistry Generic Elective and Programme (Semester-III)
Comme	Orthogram
Course	Outcomes
	After completion of these courses students should be able
GE-3 /C-1C	CO-17. To learn about chemical
	thermodynamics and its application
	CO-18. To learn equilibrium for chemical reactions
	CO-19. To learn about equilibrium in ionic solutions
	CO-20. To learn aromatic hydrocarbon
	CO-21. To learn about organometallic compounds
	CO-22. To learn about aryl halides
	CO-23. To study about alcohols, phenols, ethers and carbonyl
	compounds
	CO-24. To learn determination of pH of various solutions
	CO-25. To identify pure organic compounds
SEC-1 Basic Analytical Chemistry	CO-26. To learn about fundamental mathematical procedure
	and their applications in chemistry
	CO-27. To learn computer programming for statistical analysis
	CO-28. To handle numeric data
	Co-29. To learn about application of basic analytical procedures
	in chemistry
Course Outcomes B. Sc Ch	emistry Generic Elective and Programme (Semester-IV)
Course	Outcomer
Course	Outcomes
	After completion of these courses students should be able
GE-4 /CC-1D	CO-30. To learn about the Carboxylic Acids and Their
	Derivatives
	CO-31. To study about Amines and Diazonium Salts
	CO-32. To learn amino acids
	CO-33 To learn about aromatic nitro compounds.
	CO-34. To learn carbohydrate chemistry
	CO-35. To learn about general group trends in periodic table
	CO-36. To study Lanthanides and actinides
	CO-37. To learn Coordination chemistry
	CO-38. To study about Industrial chemistry
	CO-39. To study Error Analysis and Computer Applications

SEC-2 Pharmaceutical Chemistry	CO-40. To learn about pharmaceuticals chemistry including drugs	
	CO-41. To study about fermentation	
<u>Course Outcom</u>	es B. Sc Chemistry Programme (Semester-V)	
DSE-1A Green Chemistry	CO-42. To know about the principles of Green Chemistry and	
	Designing a chemical synthesis	
	CO-43. To study some examples of Green Synthesis/	
	CO 44 Te learn about Future Trands in Crean Chamistery	
SEC 2 IT Shill for Character	CO-44. To learn about Future Trends in Green Chemistry	
SEC-3 11 Skill for Chemists	techniques and measurement	
	CO-46. To study Algebraic operations. Differential calculus	
	and Numerical integration	
	CO_{-47} To know about basics of Computer programming	
	CO-48 Acquire Practical Knowledge on Handling numeric	
	data Numeric modelling and tatistical analysis	
Course Outcom	as R Sc. Chemistry Programme (Semester-VI)	
Course Outcomes D. Sc Unemistry Programme (Semester-v1)		
DSE-1B Polymer Chemistry	CO-49. State the basic concept of polymer.	
	CO-50. Relate Tm, Tg and its significance.	
	CO-51. Apply the Polymerization techniques and Polymer	
	CO-52. Differentiate Natural and synthetic rubbers.	
	CO-53. Distinguish Thermoplastic and thermosetting resins.	
SEC-4 Analytical Chemical	CO-54. To learn the basic concept of carbohydrates, protein,	
Biochemistry	enzymes, lipids etc.	
	CO-55. To acquire knowledge about the diagnostic approach	
	of blood and urine analysis.	
	CO-56. To gather hands on laboratory experience about	
	estimation of carbohydrates, lipids and proteins.	
	CO-57. To acquire hands on experience on isolation of	
	protein, determination of cholesterol and nucleic acids etc.	
	CO-58. To develop basic knowledge about data handling using	
	MS Word, MS Excel and MS PowerPoint.	